Organorhodium Complexes of cis- and trans-Divinylcyclopropane; The Crystal Structure of (\pm) -ZZ-1,2,3- η -5,6,7- η -Heptadienediylrhodium(1) Hexafluoro-acetylacetonate

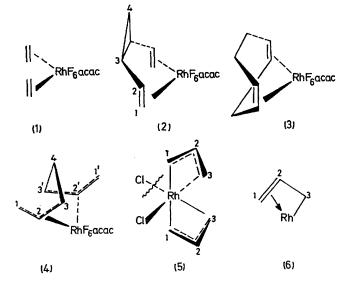
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Summary Both cis- and trans-divinylcyclopropane form 1:1 hexafluoroacetylacetonatorhodium(I) complexes; in the former, the integrity of the cyclopropane ring is maintained whilst in the latter it is essentially opened, as shown by a crystal structure determination.

Cis-DIVINVLCYCLOPROPANE, prepared as previously described,¹ reacts at -20° in iso-pentane solution with bisethylenehexafluoroacetylacetonatorhodium(I) (1) to give a yellow crystalline 1:1 complex by ethylene displacement, m.p. 79-80°, ¹H n.m.r. ($C_{6}D_{6}$), δ 5.95 (acac C-H) 4.34 [H(2)] 3.33 [H(1)_{cis}, J_{cis} 8 Hz] 2.16 [H(1)_{trane}, J_{trane} 13 Hz] 1.02 $[H(4)_{eyn}]$ and 0.62 $[H(2) H(4)_{entil}]$. The vanishingly small coupling between H(2) and H(3) indicates a transoid configuration (2). ¹³C n.m.r. chemical shifts and ¹⁰³R-¹³C couplings are typical of vinyl-olefin-rhodium complexes, and the high-field uncoupled C(3) and C(4) resonances demonstrate the integrity of the cyclopropane ring; δ (C₆D₆) 90.2 (acac-C) 81.8 [C(1)], 63.1 [C(2)], 22.7 [C(3)], and 11.4 [C(4)]. As anticipated,² (2) is stable in solution at room temperature. At 65° in C₆D₆, conversion into a new complex occurs with < 10% side-products at completion ($t_i = 10$ h). Structure (3) is indicated for the rearrangement product, as treatment of

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the resulting solution with [2H6]-dimethyl sulphoxide, causes quantitative degradation to cyclohepta-1,4-diene. Reaction



probably proceeds by trace dissociation of (2) to cisdivinylcyclopropanet followed by rearrangement and recombination, since it is suppressed by addition of 20 mole % of (1).

Trans-divinylcyclopropane, prepared by reacting methylenetriphenylphosphorane with trans-1-formyl-2-vinylcyclopropane,³ displaces ethylene from (1) at room temperature, again giving a monomeric 1:1 complex, m.p. 68-69°, ¹H n.m.r. (C_6D_6) , δ 5.97(acac C-H) 4.50 [H(1)_{eis}, H(2)] 4.08 [H(1)_{irans}, J_{trans} 11.4 Hz) 1.93 [H(3), $J_{2,3}$ 6.1 Hz] and 0.80 [H(4), $J_{3,4}$ 5.7 Hz]. Whilst the equivalence of the H(4) protons suggests C_2 symmetry there are several compatible alternative structures. ¹³C n.m.r. chemical shifts and ¹⁰³Rh-¹³C couplings accord better with a bis-allylic form (4) in solution, δ (C₆D₆) 115.3 [C(2), $J_{\text{Bh-C}}$ 6 Hz] 89.5 (acac-C) 56·4 [C(1), J_{Rh-C} 9 Hz] 30·9 [C(3), J_{Rh-C} 7 Hz] and $24 \cdot 4$ [C(4)] with assignments confirmed by off-centre double resonance. The crystal structure of (4) confirms the bis- π -allyl configuration⁴ in the solid state, and also shows it to be the ZZ-configurational isomer. The complex (4) is thermally stable $(20 \text{ h}, 100^\circ)$. In the presence of a catalytic quantity of (1), however, trans-divinylcyclopropane is slowly converted at 95° in C_6D_6 into cyclohepta-1,4-diene.

Crystal data: RhC₁₂H₁₁F₆O₂, orthorhombic, a = 8.599(1), b = 14.684(3), c = 11.264(2) Å, Z = 4, space group C222₁, 521 observed reflections collected on a Syntex P21 diffractometer. The structure was solved by the heavy-atom method and refined to a current R value of 0.069 (all atoms with anisotropic temperature factors); data quality is limited

by crystal decomposition, and the accuracy of the refined structure by high correlations due to pseudo-symmetry and by disorder in the CF₃ groups; the molecule is shown in the Figure.

FIGURE. trans-Divinylcyclopropanerhodium(1) hexafluoroacetylacetone complex. Only one set of disordered fluorine atoms is shown. Dihedral angle $C(1)C(2)C(3)/C(2)C(3)C(4) = 38^{\circ}$.

The cyclopropane ring has opened and the overall structure resembles that of bis-η-allylrhodium chloride dimer, (5).⁵ However, the bis- η -allyl unit in (4) is distorted from octahedral geometry (Figure) and the effect of the bridging atom C(4) is to increase the extent of η -allyl localisation, as in (6), with partial rehybridisation of C(3).

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- ^t The half-life for rearrangement of *cis*-divinylcyclopropane at 65° is approximately 5 s.

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